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1000 mL. The molarity of a
solution is a ratio of the
moles of solute per liters
of solution. The units for
molarity are written as
mol/L or M. This measurement
is used to perform
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molsolute L. solution 1 L-
1000 ml. A solution can be
made less concentrated in a
process called dilution.
This is accomplished by
adding more solvent.

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moles of.

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Worksheet Solve the
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How many grams of silver
chromate will precipitate
when 150. mL of 0.500 M
silver nitrate are added to
100. mL of 0.400 M potassium
chromate? $2 \text{ AgNO}_3(\text{aq}) + \text{K}_2\text{CrO}_4(\text{aq}) \rightarrow \text{Ag}_2\text{CrO}_4(\text{s}) + 2 \text{ KNO}_3(\text{aq})$
0.150 L AgNO_3
0.500 moles AgNO_3 1 moles

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1 L =

1000 mL. The molarity of a

solution is a ratio of the

moles of solute per liters

of solution. The units for

molarity are written as

mol/L or M. This measurement

is

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As we learned previously,

double replacement reactions

involve the reaction between

ionic compounds in solution

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and, in the course of the reaction, the ions in the two reacting compounds are "switched" (they replace each other). Because these reactions occur in aqueous solution, we can use the concept of molarity to directly calculate the number of moles of reactants or products that will ...

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can be produced from the
reaction of 2.50 L of 0.250
M Calcium chloride with an
excess of phosphoric acid?

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magnificent port to an uncertain destiny. This was a time when soldiers and sailors on their final flings jammed the Times Square movie houses featuring lavish stage shows as well as the nightclubs like the Latin Quarter and the Copacabana; a time when bobby-soxers swooned at the Paramount over Frank Sinatra, a sexy, skinny substitute for the boys who had gone to war. Richard Goldstein's Helluva Town is a kaleidoscopic and compelling social history that captures the youthful electricity of wartime and recounts the important role New York played in the

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national war effort. This is a book that will prove irresistible to anyone who loves New York and its relentlessly fascinating saga. Wartime Broadway lives again in these pages through the plays of Lillian Hellman, Robert Sherwood, Maxwell Anderson, and John Steinbeck championing the democratic cause; Irving Berlin's *This Is the Army* and Moss Hart's *Winged Victory* with their all-servicemen casts; Rodgers and Hammerstein's *Oklahoma!* hailing American optimism; the Leonard Bernstein-Jerome Robbins production of *On the Town*; and the *Stage Door Canteen*. And these were the

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days when the Brooklyn Navy Yard turned out battleships and aircraft carriers, when troopships bound for Europe departed from the great Manhattan piers where glamorous ocean liners once docked, where the most beautiful liner of them all, the Normandie, caught fire and capsized during its conversion to a troopship. Here, too, is an unseen New York: physicists who fled Hitler's Europe spawning the atomic bomb, the FBI chasing after Nazi spies, the Navy enlisting the Mafia to safeguard the port against sabotage, British agents mounting a vast intelligence operation. This is the city

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that served as a magnet for European artists and intellectuals, whose creative presence contributed mightily to New York's boisterous cosmopolitanism. Long before 9/11, New York felt vulnerable to a foreign foe. Helluva Town recalls how 400,000 New Yorkers served as air-raid wardens while antiaircraft guns ringed the city in anticipation of a German bombing raid. Finally, this is the story of New York's emergence as the power and glory of the world stage in the wake of V-J Day, underlined when the newly created United Nations arose beside the East River,

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