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CAPACITY $2 \frac{1}{T} = \frac{\partial S}{\partial U} \quad (7) =$
 $\frac{1}{T} \frac{\partial S}{\partial q} \quad (8) = k [\ln(q+N) + 1 \ln q$
 $1] \quad (9) = k \ln \frac{q+N}{q} \quad (10) = k \ln$
 $\frac{U+N}{U} \quad (11) \quad T = k \ln \frac{U+N}{U} \quad (12)$

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We can invert this to obtain $U(T)$:

$$e = kT = U + N U \quad (13) \quad U = N e$$

$$= kT \quad (14) \quad \text{Finally, the heat}$$

capacity is obtained from $C = \left(\frac{\partial U}{\partial T} \right)$

$$\left(\frac{\partial U}{\partial T} \right) = N e = kT \quad (15) \quad e = kT \quad (16)$$

$$= 2N e = kT \quad (17)$$

$$\left(\frac{\partial U}{\partial T} \right) = 2N e = kT$$

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Step 1 of 7. (a) Use the values of
Planck ' s constant and the speed of
the light to find the value of λ .

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The Planck ' s constant is and the speed of light is . Thus, the value of h in is, Round off to three significant figures, the value of h in is . Therefore, the given result is proved. Comment (0) Step 2 of 7.

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